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Chapter 22 22Å'1 Chapter 22 Chemistry of The NonMetals â€œ Descriptive chemistry of the elements is consistent with the various principles discussed earlier.

Chapter 22 Chemistry of The NonMetals 22.1 General

Discussions of the chemistry of the nonmetals therefore tend to focus on the following elements: H, C, N, O, F, P, S, Cl, Se, Br, I, and Xe. One way of visualizing the difference between metals, semimetals or metalloids, and nonmetals is the plot of average valence electron energies (AVEE) in Figure N.2.

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CHEMISTRY FACULTY: Paul Bonvallet, Chair Judith Amburgey-Peters (on leave Fall 2016) Aaron Baker Paul Edmiston Karl Feierabend Spring Knapp Melissa Mullen-Davis Michael Peterson Eric Popczum Mark Snider Sarah Sobeck James West (on leave 2016-2017) Chemistry is broadly defined as the study of the properties of matter and how matter is transformed.

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This book is a new attempt to interrelate the chemistry of the non-metals. In the early chapters, simple compounds of the non-metals with the halogens, hydrogen, and oxygen are surveyed, permitting a large area of chemistry to be discussed without the burden of too many facts.

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comes from fact that phosphoacylglycerols are lipids that contain a phosphate group. - Hydrophobic effect: "Hydrophobic lipid tails avoid water but associate with each other, resulting in phospholipid bilayer" (shown below) (Pg. 9. of the summer 2006 Thinkbook by Dr. S. Hardinger).

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1.5 Metals & Non-metals 14 1.6 Chemical symbols 15 1.7 Compound 15 1.8 Mixture 15 1.9 Physical separation 16 1.10 Physical and chemical change 16 1.11 Elements, mixtures or compounds? 17 1.12 Atomic theory 17 ... Fundamentals of Chemistry ...

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